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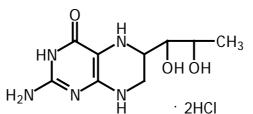
DATA SHEET

TETRAHYDROBIOPTERIN

Synonym: 2-amino-6-[(1R,2S)-1,2-dihydroxypropyl]-5,6,7,8-tetrahydro-4(1H)-pteridinone

Please note that none of the products shown below are sterile. They are sold in ampoules for laboratory use only.

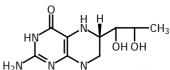
CAS No. 17528-72-2



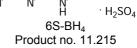
6R,S-BH₄ Product no. 11.209

CAS No. 69056-38-8





CAS No. 109784-74-9



oduct Name Pr

Product Name	Prod. No.	Abbreviation	Molecular Formula	M.W.
(6R,S)-5,6,7,8-Tetrahydro-L-biopterin dihydrochloride	11.209	6R,S-BH₄·2HCl	C ₉ H ₁₅ N₅O ₃ · 2HCl	314.2
(6R)-5,6,7,8-Tetrahydro-L-biopterin dihydrochloride	11.212	6R-BH₄·2HCI	C ₉ H ₁₅ N₅O ₃ · 2HCl	314.2
(6S)-5,6,7,8-Tetrahydro-L-biopterin sulfate	11.215	6S-BH ₄ ·H ₂ SO ₄	$C_9H_{15}N_5O_3\cdot H_2SO_4$	339.3

Chemical reduction of biopterin yields two diastereoisomers, 6R- and 6S-BH₄. 6R-BH₄ is the natural form of tetrahydrobiopterin. The 6R,S-BH₄ that we manufacture contains about 70% of 6R-BH₄ and 30% of 6S-BH₄.

Description	White to light yellow	powder		
Biochemical functions	Tetrahydrobiopterin is a natural occurring cofactor of the aromatic amino acid hydroxylases and is involved in the synthesis of tyrosine and the neurotransmitters dopamine and serotonin. It is also essential for nitric oxide synthase catalysed oxidation of L-arginine to L- citrullin and nitric oxide. Tetrahydrobiopterin is involved in many other biochemical functions, many of which have been just recently discovered.			
Solubility	6R- and 6R,S-BH ₄ ·2HCI are freely soluble in water. The solubility of 6S-BH ₄ ·H ₂ SO ₄ is 2.3 g per 100 g of water (22°C). A 1 mM solution of 6R-BH ₄ in water gives a pH of 3.0 and a 1 M solution of 6R-BH ₄ in water gives a pH of 0.45.			
Analytical methods	HPLC conditions:	column: eluant: flow rate: wavelength: solution:	Whatman Partisil 10 SCX 30 mM NaH ₂ PO ₄ , pH 3 1.5 ml/min 254 nm 160 mg BH ₄ / 100 g H ₂ 0 with degassed ascorbic acid solution (0.16%)	

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Analytical methods (contd.)	UV ε ₂₆₇	16 x 10 ³ M ⁻¹ cr	n ⁻¹ in 0.1N HCl			
	TLC: BH ₄ solutions are not	stable in the presenc	e of air to perform TLC.			
Purity	Product no. 11.212 is > 99.5% pure, contains less than 0.1% of $6S-BH_4$ and has no sodium chloride.					
Stability	 Tetrahydrobiopterin is very hygroscopic. It reacts with oxygen especially in neutral and alkaline solutions. Due to oxidation tetrahydrobiopterin solutions become yellow but at -20°C or colder, solutions are relatively stable. 0.1 mM solutions of tetrahydrobiopterin in water are much less stable than 1 mM solutions, i.e. after 1 hour open at room temperature, 0.1 mM solutions degrade by about 25% whereas 1 mM solutions degrade only by 2%. After 3 hours, 0.1 mM solutions degrade by more than 60% and 1 mM solutions degrade only by 10%. Dry powder has different stability depending on conditions, as described in the table below. 					
	In ampoules	RT several months	-20°C several years			
Storage	Storage conditions are not controlled during shipment and we cannot guarantee that the customer will receive the shipment within 6 weeks. There is no cause for concern when ampoules are delayed in the post as they are stable for several months at room temperature. Customers who have purchased 6R-BH ₄ in the past can estimate the time required for shipment.					
	at -20°C or colder. After an ampoule has and then it can be stored in a freezer. in oxygen free water with the lowest order to increase the stability of BH_4 may be added.					
Safety information	Tetrahydrobiopterin is known to be safe for the intended use. Avoid prolonged inhalation of the fine dust of tetrahydrobiopterin which is very acidic. Otherwise there are no special precautions required in handling this product.					
References	Biochem., <u>2</u> , (1991), 411-423. Tetrahydrobiopterin deficiency: Claus W. Heizmann and Jean- New Tetrahydrobiopterin-Depe (1993), 261-286.	From phenotype to g Louis Dhondt, Pteridir ndent Systems. Seyn iopterin in biological ti	vid S. Duch and Gary K. Smith, J. Nutr. enotype. Nenad Blau, Beat Thöny, nes, <u>4</u> , (1993), 1-10. nour Kaufmann, Annu. Rev. Nutr., <u>133</u> , ssues and fluids. Fukushima, T. and J.			

Further data sheets can be found on our website www.schircks.ch

The information in this publication is based on our current knowledge and experience. It does not relieve users or processors from carrying out their own precautions and tests.